

Saturated Unsaturated Supersaturated Graph Answers

Unsaturated, Saturated, and Supersaturated Examples. Check Your Answer The liquid inside reusable hand warmers will crystalize over time. However, they can be reheated in the microwave and go back to the liquid form. Explain why the crystals form and why reheating the hand warmers "removes" the crystals. Check Your Answer.

Types of Solutions: Saturated, Supersaturated, or Unsaturated

A graph can illustrate what solution is saturated and unsaturated. If the point is on the line, then the solution is saturated, while if it is below the line, the solution is unsaturated. share with ...

What is saturated and unsaturated solutions - answers.com

Best Answer: The solution will be saturated when the grams of solute dissolved = the solubility of the salt. In other words the solution will be saturated at the line on the graph. In other words the solution will be saturated at the line on the graph.

Chemistry: How can you tell if the solution is saturated ...

A graph can illustrate what solution is saturated and unsaturated. If the point is on the line, then the solution is saturated, while if it is below the line, the solution is unsaturated.

How does a graph show a saturated solution ... - answers.com

Chemistry question: saturated, unsaturated or supersaturated? Using a graph, how do you predict whether certain solutions would be considered saturated, unsaturated, or supersaturated??? If it helps, the solutions are 75 g KNO₃/100g H₂O at 40 degrees C and 60 g KNO₃/100g H₂O at 40 degrees C.

Chemistry question: saturated, unsaturated or ...

To answer this, find where the curve of KCl crosses 30°C. I have marked this point A on the graph. You then read across and determine the grams of solute which is 37 g. Solubility Curve Practice Problems Worksheet. For most substances, solubility increases as temperature increases.

Solubility Curve Practice Problems Worksheet 1

In a nutshell, unsaturated solutions have points below the line on the graph, saturated is on the line and supersaturated is above the line on the graph.

Solubility Curves - Saturated, Unsaturated, Supersaturated Solutions

approximately 26 more grams [the solubility of potassium chlorate is approximately 30g at 65 Celcius] 10. A solution of potassium iodide at 70°C contains 200 g of dissolved solute in 100 cm³ water. The solution is allowed to cool.

KEY Name, Date, Hr/Per Solubility Curves

Chemistry saturated, unsaturated or supersaturated question.? So im stuck on a question At 90 degrees celsius, 10g of potassium chlorate dissolved in 100g of water is the solution saturated, unsaturated or supersaturated and how do you know so i know what the 3 terms means, i just dont know a way o figure it out with the above info.

Chemistry saturated, unsaturated or supersaturated ...

Best Answer: How are you using the term "supersaturated"? If that is used properly, it is a true solution containing no solid but containing a concentration of solute greater than a saturated solution. If you are using it to mean a saturated solution in contact with excess solid, then two of the solutions are supersaturated.

Chemistry question: unsaturated, saturated, or ...

When 20 grams of potassium chlorate, KClO₃, is dissolved in 100 grams of water at 80 °C, the solution can be correctly described as: unsaturated. saturated. supersaturated. When 30 grams of potassium nitrate, KNO₃, is dissolved in 100 grams of water at 20 °C, the solution can be correctly

described as: unsaturated.

Solubility Diagram - ScienceGeek.net Homepage

What is an saturated and unsaturated solution? Update Cancel. ... Now after this extent Further increase in temperature will also not dissolve any solute.. then the solution is known as super saturated solution. ... Answered Jun 2, 2018 · Author has 704 answers and 72.8k answer views.

What is an saturated and unsaturated solution? - Quora

Solubility is referred to in three different terms: saturated, unsaturated and supersaturated. A saturated solution is a solution that contains the maximum amount of dissolved solute.

Solubility and Solubility Curves - Video & Lesson ...

Use the provided solubility graph to answer the following questions: For questions 1 - 4 an amount of solute is given, and a temperature is stated. If all of the solute could be dissolved in 100 g of water at the given temperature, would the resulting solution be unsaturated, saturated, or supersaturated? 1. 60 g KCl at 70 °C _____ 2. 10 g KClO

Worksheet: Solubility Graphs Name

Values on the graph below a curve represent unsaturated solutions - more solute could be dissolved at that temperature. Values above a curve represent supersaturated solutions, a solution which holds more solute that can normally dissolve in that volume of solvent.

3.1 Solubility Curves - Chemistry LibreTexts

I'm taking an online chemistry course, and it doesn't explain this concept very well. I am confused on how to identify whether a solution on the graph is unsaturated, saturated, or supersaturated. I added the picture of the graph that goes along with the questions. Please explain how you got your answer, I do not want just the answers.

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